PATENT ABSTRACTS OF JAPAN

(11) Publication number:

61-004232

(43) Date of publication of application: 10.01.1986

(51) Int. CI.

H01L 21/304

(21) Application number: 59-125760

(71) Applicant: NEC CORP

(22) Date of filing:

19. 06. 1984

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(54) CLEANING METHOD OF SEMICONDUCTOR SUBSTRATE

(57) Abstract:

PURPOSE: To safely and readily remove any station on the surface of a silicon wafer, by dipping a semiconductor substrate in a solution of an organic acid being bubbled by ozone or oxygen and thereby treating the substrate.

CONSTITUTION: An organic acid (e.g., formic acid or acetic acid) is filled into a cleaning tank and heated (to 100W150°C). A semiconductor substrate is dipped in this liquid, and ozone or oxygen is supplied from the bottom of the tank so as to bubble the liquid, whereby the substrate is cleaned by the bubbles. Any heavy metal on the wafer forms a formate or an acetate, and any organic contaminant is decomposed by ozone, whereby stains on the surface of the substrate can readily be cleaned out.

LEGAL STATUS

[Date of request for examination]

[Date of sending the examiner's decision of rejection]

[Kind of final disposal of application other than the examiner's decision of rejection or application converted registration]

[Date of final disposal for application]

[Patent number]

[Date of registration]

[Number of appeal against examiner's decision of rejection]

[Date of requesting appeal against examiner's decision of rejection]

[Date of extinction of right]

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⑩日本国特許疗(JP)

10 特許出願公開

· ❷公開特許公報(A)

昭61 ~ 4232

@Int, CI;

識別配号

广内移现番号

砂公開 昭和61年(1986)1月10日

H 01 L 21/304

D-7131-5F

審査請求 未請求 発明の数 1 (全 2 頁)

の発明の名称

半導体基板の洗浄方法

创特 翻 昭59-125760

2011 爾 昭59(1984)6月19日

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1. 発明の名称

申導体基板の施律方法

2.特許健康の範囲

(1)半導体基項を有機酸化オゾンノ酸素をペプル した鬱æ中に及波し、処理することを辞録とする 単導体基根の洗浄方法。

3.発明の詳細な説明

〔技切分野〕

. 本発明は半単体基なの洗浄方法に関するらので

〔従来改術とその問題点〕

* LSI 製造において、ウェハの疣が工程は各プロ セス間にゴミ、行れを除去する双多くの処理値数 を必要とし、製品の少額を向上に重要な铰距とな つている。との洗浄工程で使われている主な旅と してはアルカリ長と敷系化大別される。公知の説 静蔵としてはアンモニアホー過便化水準系、又は 塩酸-過酸化水沸茶(適称 RCA 洗浄ともいう)。 さられ、餌食ー過酸化水常果、碗酸-硝酸系等の

混戯容波が用いられている。資液は通常60~ 150 ℃に加弱され、時には超音波を併用することもあ

しかしながら、アンモニナー過酸化水素系や塩 酸-過度化水素系では、過酸化水米がすぐに肌の と発生期のOに分解してしまうので、酸化力が劣 化し、波净力が拘続せず間欠的に過酸化水泵を補 充する必要があり、工程としては業績になる。袋 酸・過酸化水泉系も同じ産由で不利である。との 征使一過酸化水煮果や硫酸-硝酸系等は原波とし て処理する場合の問題がある。それは既敢を勇士 する場合、水と反応して100 で以上なり安全上危 険がむり、このため大量の水を必要とし、又朔線 を使う場合は、温素酸化物等が発生し公告上規劃 される等、大量に使うことはできないという不利 な点がある。すなわち、処理費用が高くなる。

又上記の最一追像化水素系に代る処理方法とし ての公知例として特公昭 52-012083 号の半導体器 桜のレジスト院除去方法がある。この方法は硫酸 を用いて過酸化水果の代きにオゾンを用いる方法

詩國昭61-4232(2)

すなわち、本発明は半導体器被(タエハ)上の

黑金属性 学敏塩、又は酢酸塩を作り有級污染物質

はオノンで分解するととにより、発神を行わりと

以下に本発明の一典的例について説明する。

本発明は洗浄槽化有機酸(例えばギ酸、酢酸)

を入れ、これを加温(100 ~ 150 ℃)し、この策

内に中導体務板を投資し、糖底部よりオゾンノ酸

果をパソルして放益板比おてて先浄を行うもので

本発明で用いた洗浄物はテフロン製のものを用。

い、液混は投込型石英ヒーメーで加勢・飼御した。

又オソンノ酸寒は糖底部よりパイレックス製ガス

分配能によりパプルして行つた。何型機を用いて

徒来の RCA 売砂波の N元 OH - 馬 O₂ - H₂O₃ (1 : 1 :

さらに本発明による方法では、 前配の周 じ槽を

丹いて同一ロットのフェハを用いて敵ウェハをポ

5) 7、被选-85℃、10分圆壳净仓行口火。

次に本発明と従来例とを比較しながら説明する。

するものである。

ある.

であるが、この方法では弱敗を廃议処理するのに 大量の水(~10倍飛択)を必要とする等、安全上、 費用の点で問題があつた。

(発明の目的)

本 為明 以 安全上 を 題 め、 か つ 低 魔 な 処理 を 行 な り と と が て き る よ う 化 し た 処理 万 战 を 役 供 す る も の で ある。

[残弱の構成]

本発明性有機酸を洗冷数として、 週内化オゾン ノ陰楽をパブルしてシリコンクエハの表面汚れを 験去することを解散とする半導体基板の洗浄方法 である。

[発明の無理・作用]

有根酸のうち半段は強酸で、湿定性をもつている。一方、雌酸は弱酸であり、それぞれ類金類(M)と反応して-COOH 左のBを置換し、-COGM なる金属銀を作る。

又オソンをパブルすることにより、有機汚染物 質をオゾンにより酸化し、分解除安することがで きる。

ツトパント選圧) は従来法のものと大きな違いはな

以上のように本発明の洗浄方法は従来法との洗 浄効果と比較して同等以上であり、死分に目的が 遠成される。

帮許也順人 日本包気株式会社

代德人 非理士 管 野

取又は酢酸溶液に浸漬する。とのとき、この容様 も 160 ~ 160 ℃に加温する。るらに独庭がよりオ ソンノ酸素をパブルして~10分筒洗浄を行つた。

世来被のNFLOH-LLO₂-R₂O系溶散で洗浄したクエハと、本発明によるギ酸ーオソン・O4系確認で洗浄したクエハとをステーム処理(950 ℃、10分 R₂-O₂学成気)を行い、非緩触法によるライフタイムを測定した結果と、1606 ℃-V 法により評価したV_{F3}(フラットパンド電圧)とを表-1だ示す。

-

微多方法	ライフタイム (#500)	VPB(V)
從来波(NHLOR系)	7	~0.9
本発明の方法(オゾンノギ酸)	20	-1-0

[発明の効果]

第1級に示したよりに太勢明による沈浄方法は 従来の BCA 洗浄核に比べて、再結合ライフタイム の娘が~3倍となり、洗浄効果が優れている事を 示している。又MOS C-V法より求めた Ves (フラ



く、-1.67 であつた。

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(19) Japanese Patent Office (JP) (12) Official Gazette for Unexamined Patent Applications (A)

(11) Japanese Patent Application Kokai Publication No. S61-004232

(43) Publication Date:

January 10, 1986

Number of Claimed Inventions:

1 (total of 2 pages)

Request for Examination:

Not requested

(51) International Class.⁴ Identification No.

JPO File No.

H 01 L 21/304

D-7131-5F

(54) Name of Invention:

Cleaning Method for Semiconductor Substrate

(21) Application No.:

S59-125760

(22) Application Date:

June 19, 1984

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Specifications

Name of Invention

Cleaning Method for Semiconductor Substrate

- Claims 2.
- (1) A cleaning method for a semiconductor substrate is characterized by a process in which a semiconductor substrate is dipped into an organic acid that is fed with bubbles containing ozone/oxygen.

3. " Detailed Description of the Invention

<Technical Field>

This invention pertains to a cleaning method for a semiconductor substrate.

<Prior Art & Related Problems>.

During LSI production, wafers must be cleaned several times between processes in order to remove dust and dirt, and a technique for doing so has become important from the standpoint of improving production yields. The primary solutions used in such a cleaning process are generally classified into alkali and acid type solutions. Examples of well-known cleaning solutions include aqueous ammonia – hydrogen peroxide and hydrochloric acid – hydrogen peroxide (also commonly referred to as RCA cleaning) types. In addition, mixed acid solutions are used such as sulfuric acid – hydrogen peroxide and sulfuric acid – nitric acid types. Solutions are normally heated to a temperature of 60 – 150°C and sometimes undergo ultrasonic wave treatment during heating.

However, in the cases of ammonia – hydrogen peroxide and hydrochloric acid – hydrogen peroxide types, since the hydrogen peroxide quickly breaks down into H₂O and nascent-state O, the deterioration in oxide strength causes inconsistent cleaning strength. Therefore, it becomes necessary to periodically replenish the solution with hydrogen peroxide, resulting in a complicated process. Sulfuric acid – hydrogen peroxide type solutions are also unfavorable for the same reason. Problems exist in cases where these sulfuric acid – hydrogen peroxide and sulfuric acid – nitric acid types are used as waste solutions during treatment. In cases where sulfuric acid is discarded, there is a safety concern when it reacts with water and reaches a temperature of 100°C or more. For this reason, a large volume of water is needed. Furthermore, when sulfuric acid is used, nitrogen oxides are generated which need to be regulated due to pollution standards, and for this reason, sulfuric acid cannot be used in large amounts. In short, the problems noted above result in high treatment costs.

In addition, Patent No. S52-012063 describes a method for removing resist films from semiconductor substrates in a well-known treatment method that takes the place of the aforementioned acid – hydrogen peroxide type solutions. According to this method, sulfuric acid is used along with ozone, which takes the place

of hydrogen peroxide. However, according to this method, a large volume of water (up to a tenfold dilution) is required in order to discard the sulfuric acid, which presents a problem from the standpoint of safety as well as cost.

<Purpose of the Invention>

The purpose of this invention is to present a treatment method in which it is possible to maintain a high level of safety as well as a low cost.

<Constitution of the Invention>

The method for cleaning semiconductor substrates as described in this invention is characterized by a process in which an organic acid is used as a cleaning solution and bubbles containing ozone/oxygen are fed into the solution tank in order to remove dirt from the surface of silicon wafers.

<Principle/Operation of the Invention>

Among the various types of organic acids, formic acid is a strong acid that contains reduction properties. On the other hand, acetic acid is a weak acid in which the hydrogen within the -COOH group becomes displaced once the acetic acid reacts with the various heavy metals (M), resulting in the formation of a -COOH metallic salt.

Furthermore, by feeding bubbles containing ozone into the solution tank, organic contaminants become oxidized by the ozone, making it possible to remove these contaminants through decomposition.

In other words, this invention seeks to conduct a cleaning operation in which heavy metals located on top of semiconductor substrates (wafers) create formic acid salt or acetic acid salt, and organic contaminants are decomposed through the use of ozone.

<Embodiments>

The following is an explanation of this invention in which a prior example is used for comparison.

First, an embodiment of this invention will be explained.

According to this invention, the cleaning operation is conducted as follows. A cleaning tank is filled with an organic acid (such as formic acid, acetic acid, etc.), and the solution is heated (to a temperature of 100 – 150°C), after which a semiconductor substrate is dipped into the solution. Ozone/oxygen bubbles are fed from the bottom of the tank and clean the substrate as they hit up against the surface.

The tank used in this invention is made of Teflon, and the solution temperature is controlled through the use of an immersion type quartz heater. The ozone/oxygen bubbles are fed into the tank from the bottom using a gas distributor made of Pyrex. The same type of tank is used to conduct cleaning with a prior form of RCA cleaning solution comprised of $NH_4OH - H_2O_2 - H_2O$ (1:1:5) at a solution temperature of 80°C for 10 minutes.

In addition to using the same tank described above, the method of this invention calls for the use of a wafer from the same lot, and said wafer is to be dipped into a solution of formic acid or acetic acid. While the wafer is within the tank, the solution is heated to a temperature of 100 - 150°C. In addition, ozone/oxygen bubbles are fed through the bottom of the tank so that cleaning can be conducted for a period of 10 minutes.

The wafer that is cleaned using the $NH_4OH - H_2O_2 - H_2O$ solution according to the prior method, and the wafer that is cleaned using the formic acid – ozone – O_2 solution according to this invention both undergo a steam treatment (in an environment of $H_2 - O_2$ at 950°C for 10 minutes), after which lifetime measurements are conducted using a non-contact method. These results are shown in Table 1 along with the V_{FB} (Flat Band Voltage) figures, which are evaluated using the MOS C-V method.

Table 1

Cleaning Method	Lifetime (µ sec)	V _{FB} (V)
Prior Method (NH ₄ OH)	7	-0.9
Method of this Invention	20	-1.0
(Ozone/Formic Acid)		

<Effect of the Invention>

As shown in Table 1, the cleaning method described in this invention has a recombination lifetime value that is approximately three times that of the RCA cleaning solution from the prior method, which indicates a superior cleaning effect. In addition, the V_{FB} (Flat Band Voltage) figure obtained through the MOS C-V method shows a level of -1.0V, which is not significantly different from the result shown for the prior method.

Based on the above, the cleaning method described in this invention provides a cleaning effect that is

equal to or greater than that of the prior method, thus sufficiently achieving the purpose stated above.

In addition, the problem noted above with respect to the generation of heat due to the reaction between

water and sulfuric acid or nitric acid is prevented, thus this invention creates very few problems from the

standpoint of pollution. This is due to the fact that formic acid and acetic acid can be easily decomposed into

CO, CO₂, H₂O, etc., at high temperatures (up to 150°C), and this decomposition is further advanced through the

use of ozone in the treatment process. As a result, this invention poses no problem with regard to discarding the

solution since it can be easily dissolved in water. This not only makes it possible to provide industrial safety as

well as a rational manufacturing operation, it also makes it possible to reduce the level of cost required to

conduct the treatment process.

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